This question paper contains 4+2 printed pages]

AG-80-2018

FACULTY OF SCIENCE

M.Sc. (Chemistry) (First Year) (First Semester) EXAMINATION OCTOBER/NOVEMBER, 2018

(CBCS Pattern)

CHEMISTRY

Paper—I (CH-412)

(Organic Chemistry)

(Wednesday, 28-11-2018)

Time: 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. := (i) Attempt All questions.

- (ii) Figures to the right indicate full marks.
- (iii) Use of logarithmic table and calculator is allowed.
- (iv) Multiple Choice Questions (MCQ) should be attempted only once on page No. 3 of answer-book with complete answer.
- 1. Attempt any three of the following:

15

- (a) Explain arenium ion mechanism with suitable examples.
- (b) Derive the Hammett equation and calculate the substituent constant of the following ethyl benzoate.

(Given data : pKa values of substituted ethyl benzoate at M-Me = 4.26, P-H = 4.20, M-NO $_2$ = 3.49, P-OMe = 4.47, M-Br = 3.80 and pK $_{aH}$ = 4.20)

Calculate:

$$\sigma_{M-Me}$$
 = ? σ_{P-H} = ? σ_{M-NO_2} = ? σ_{P-OMe} = ? σ_{M-Br} = ?

P.T.O.

- (c) Explain the effect of substrate, attacking nucleophile and leaving group in aliphatic nucleophilic substitution reaction.
- (d) What is photofries reaction? Explain the photofries reaction of anilides with suitable example.
- (e) What is Jablonski diagram? Explain in detail with suitable examples.
- 2. Attempt any three of the following:

15

- (a) Explain the generation and stability of Carbocation and Nitrene.
- (b) Explain the primary and secondary kinetic isotopic effect with examples.
- (c) Explain the term Anchimeric assistance and SNⁱ mechanism with examples.
- (d) What is Photochemistry? Explain the Norrish Type-I and II reactions with example.
- (e) Discuss the photochemistry of cis-trans isomerisation of stillbenes in the presence of photosensitiser.
- 3. (a) Comment on the following:

7

- (i) Hammond postulate
- (ii) Alternate and non-alternate hydrocarbon.

Or

What is photoreduction? Explain the photoreduction of benzophenone with mechanism.

(b) Predict the products with mechanism of the following (any four): 8

(i)
$$\frac{hv}{n \to \pi^*}$$

(ii) 2 Ph—C=O + CH₃—CH—CH₃
$$\xrightarrow{hv}$$
?
Ph

(iii) OEt
$$\frac{\text{NaNH}_2}{\text{liq. NH}_3}$$
?

$$(iv) \qquad \underbrace{\begin{array}{c} \operatorname{NO}_2 \\ \\ \\ \\ Z \end{array}} \xrightarrow{\operatorname{CN}^{\ominus}} ?$$

(Z = Electron withdrawing gap)

P.T.O.

$$(vi) \qquad \begin{array}{c} \text{OC}_2 \text{H}_5 \\ \text{Me} \\ \text{|} \\ \text{+ Ph-N-CHO} \xrightarrow{\text{POCl}_3} ? \end{array}$$

(vii) + Conc.
$$H_2SO_4$$
 \longrightarrow ? Slow ?

4. (a) Explain the Paterno-Buchi reaction and photochemistry of α, β-unsaturated ketones with examples.

Or

Discuss the following:

- (i) SNⁱ mechanism
- (ii) Benzyne intermediate reaction.
- (b) Explain the following:

8

- (i) Ortho-para ratio
- (ii) Diazonium coupling reaction.

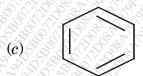
Or

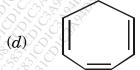
- (i) What are annulenes? Explain the aromaticity of [10] annulenes.
- (ii) Substituent constant and reaction constant.
- 5. (A) Select the *correct* answer from the given options for each of the following:
 - (i) The conversion of singlet to triplet state takes place via :
 - (a) Internal conversion
 - (b) Intersystem crossing
 - (c) Fluorescence
 - (d) Phosphorescence

- (ii) σ_x in Hammett equation indicates
 - (a) Reaction constant
 - (b) Equilibrium constant
 - (c) Rate constant
 - (d) Substituent constant
- (iii) By the absorption of correct energy hv, the electronic excitation from ground state to excited state with spin inversion gives
 - (a) Singlet state
 - (b) Triplet state
 - (c) Transition state
 - (d) None of the above
- (iv) Which of the following compounds is non-aromatic?









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- (v) A phenomenon of hyperconjugation is not shown by:
 - (a) Methyl carbocation
 - (b) Ethyl carbocation
 - (c) Isopropyl carbocation
 - (d) Tert. butyl carbocation
- (B) Write short notes on any two of the following:

10

- (i) Photodegradation of polymer
- (ii) NGP
- (iii) Taft equation.