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BR—65—2016

FACULTY OF SCIENCE

M.Sc. (Second Year) (Fourth Semester) EXAMINATION

OCTOBER/NOVEMBER, 2016

(CBCS Pattern)

INORGANIC CHEMISTRY

Paper XVII-(CH-541/1)

(Organometallic Catalysis and Fluxionality)

(Thursday, 17-11-2016)

Time : 2.00 p.m. to 5.00 p.m.

Time—Three Hours

Maximum Marks—50

N.B. :— (i) Attempt All questions.

(ii) Figures to the right indicate full marks.

1. Answer any *five* of the following :

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- Draw structure of sodium borohydride.
- Discuss the effect of porosity on catalytic effect.
- What is water gas shift reaction ?
- Explain the polymerisation of alkenes using Ziegler-Natta catalyst.
- Name the Wilkinson's catalyst and write its application.
- What are the essential properties of catalysts ?
- Highlight the applications of zeolites.
- Define and explain the term rate of fluxionality.

2. Attempt any *four* of the following :

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- Describe the use of Heterogeneous catalysis in synthetic chemistry.

P.T.O.

- (b) Explain the mechanism of Heck reactions.
- (c) Discuss the role of catalyst in Wacker process.
- (d) Discuss the principle of hydroformylation of olefins.
- (e) How stereochemical non-rigidity is detected ?
- (f) What is surface migration in relation to catalysis ?
3. Answer any *two* of the following :
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- (a) What is Epoxidation ? Explain with example.
- (b) Distinguish between chemisorption and desorption.
- (c) Draw and explain Tolmen catalytic cycle.
4. Answer any *two* of the following :
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- (a) Describe the acetic acid synthesis by Monsoanto process.
- (b) Discuss the principle of Fishcher-Tropsch synthesis
- (c) Explain the catalytic steps involved in homogeneous catalysis.
5. (A) Choose the correct alternatives :
5
- (i) In the hydrogenation of edible oil is used as a catalyst.
- (a) Roney Nickel
- (b) Devard's alloy
- (c) Palladium chloride
- (d) Vanadium pentaoxide
- (ii) The is the chemical reaction of an unsaturated halide with an alkene in the presence of a base and a palladium

catalyst to form a substituted alkene.

- (a) Hydrocarboxylation
 - (b) Heck reaction
 - (c) Birch Clemmenson's reaction
 - (d) Wolf-Kishner reaction
- (iii) Monsanto process operates at a pressure of and at to give selectivity greater than 99%.
- (a) 10–20 atm and at 50–80°C
 - (b) 60–80 atm and at 100–120°C
 - (c) 30–60 atm and at 150–200°C
 - (d) 10–20 atm and at 100–200°C
- (iv) The is an industrial method for the manufacture of acetic acid by catalytic carboxylation of methanol.
- (a) Fishcher–Tropsh synthesis
 - (b) Hydroxylation
 - (c) Wacker's process
 - (d) Monsanto process
- (v) In Wacker's process, formation of C–O double bond proceeds through with a cyclic four-membered transition state.
- (a) β -hydride addition
 - (b) β -hydride oxidation
 - (c) β -hydride elimination
 - (d) β -hydride reduction

(B) Write brief notes on (any two) :

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- (a) Fluxionality in cyclopentadienyl complexes
- (b) Reppe's catalysis
- (c) Ammonia synthesis.