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AI—140—2017

FACULTY OF SCIENCE

M.Sc. (Fourth Semester) EXAMINATION

MARCH/APRIL, 2017

(CBCS Pattern)

ORGANIC CHEMISTRY

Paper CH-542/2

(Bio-organic and Green Chemistry)

(Monday, 24-4-2017)

Time: 2.00 p.m. to 5.00 p.m.

Time— Three Hours

Maximum Marks—75

- N.B. := (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
 - (iii) Multiple choice questions (MCQs) should be attempted only once on page No. 3 of answer-book with complete answers.
- 1. Solve any three:

15

- (a) Discuss the structure of uracil and thiamine.
- (b) Give the types of RNA with their functions.
- (c) Explain the role of PTC in Darzen reaction.
- (d) Use of DMC as green reagent.
- (e) What is atom economy? Explain.
- 2. Answer the following (any three):

15

- (a) Explain the hydrolysis of Nucleic acid.
- (b) Microwave assisted Dieds Alder reaction.
- (c) Reaction specificity in enzymes.
- (d) Discuss the secondary structure of DNA.
- (e) Explain biochemical reduction with two suitable examples.

P.T.O.

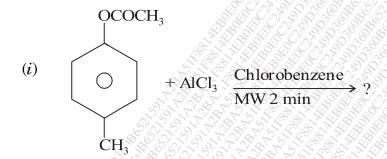
3. (A) Explain double Helix structure of DNA given by Watson and Crick. 7

Or

Why is the need of green chemistry? Explain in detail.

(B) Predict the products (Any four):

8



$$(ii) \qquad \frac{\text{Pseudomonas}}{\text{Putida}} ?$$

(iii)
$$\frac{\bigoplus_{NR_3I} \bigoplus_{NR_3I} \bigoplus_{NR$$

(iv)
$$\frac{\text{CH}_3\text{COCl[0min] Cl-AlCl}_3}{(X = 0.67) 5 \text{ min, } 0^{\circ}\text{C}} ?$$

(vii) ArRCHX + NaOCl
$$\xrightarrow{\text{CH}_3\text{CN, RT}}$$
 ?

4. (A) Discuss the twelve principles of green chemistry in detail.

7

Or

Explain Lock and Key mechanism of enzymes.

(B) Explain reactions in acidic ionic liquids and neutral ionic liquid. 8

Or

What is phase transfer catalyst? Give the advantages of PTC to green synthesis.

- 5. (A) Select the *correct* answer from the following multiple choice questions.
 - (i) following base is not present in DNA.
 - (a) Uracil

- (b) Guanine
- (c) Adenine
- (d) Cytosine

P.T.O.

WT			(4)	AI—140—201	7
	(ii)	is green protocol for organic synthesis.				
		(a)	Ultrasound	8		3
		(<i>b</i>)	Ionic liquid	59,01		3× C
		(c)	Microwave induce	d system		2
		(d)	All of the above			3
	(iii)	The termis used to describe effect of ultrasound waves in chemical reactivity.				
		(a)	Nanochemistry	(b)	Sonochemistry	
		(c)	Piezochemistry	(d)	Thermochemistry	
	(iv)	Microwaves ranges fromin wavelength in electronegative spectrum.				
		(a)	1 m - 10 m	(b)	1 cm - 1 m	
		(c)	$1 \mu m - 1 m$	(d)	1 nm – 1 cm	
	(v)	There areprinciples of green chemistry.				
	A A A B	(a)	10	(b)	12	
		(c)	14	(d)	None of these	
(B)	Writ	e shor	rt notes on (any two		1	C

- (i) Give the classification of enzyme.
- (ii) Alkylation of reactive methylene compound under solvent free condition.
- (iii) Three point attachement theory.