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AI—46—2017

FACULTY OF SCIENCE

M.Sc. (Second Year) (Fourth Semester) EXAMINATION

OCTOBER/NOVEMBER, 2017

(CBCS Pattern)

PHYSICAL CHEMISTRY

Paper XX (CH-541/3)

(Radiation Chemistry)

(Saturday, 11-11-2017)

Time : 2.00 p.m. to 5.00 p.m.

Time—3 Hours

Maximum Marks—75

N.B. :— (i) Attempt All questions.

(ii) Use of log table and calculator is allowed.

(iii) Solve MCQ's in one attempt only.

1. Solve any *three* of the following : 15

- (a) Explain characteristics of β -decay.
- (b) Explain Bethe's notation for nuclear reaction.
- (c) Explain atomic absorption coefficient.
- (d) Write a short note on hydrated electrons.
- (e) Write a note on Tunnel effect.

2. Attempt any *three* of the following : 15

- (a) Derive the expression : $N = N_0 \cdot e^{-\lambda t}$.
- (b) Explain in brief uses of Nuclear radiations.
- (c) Write a note on Linear energy transfer.
- (d) How will you explain evaporation and spallation ?
- (e) Write a note on radiolysis of water.

P.T.O.

3. Solve the following :

(a) What is fission energy ? Explain in brief reproduction factor (K). 7

Or

Discuss the stability of nucleus in terms of neutron proton ratio and binding energy. 2 g of radioactive element degraded to 0.5 g in 60 hours. In what time will it be reduced to 10% of its original amount ?

(b) Explain thermonuclear reactions in the sun and stars. 8

Or

A old wooden article shows 2.0 counts per minute per gm. A fresh sample of wood shows 15.2 count per minute per gram. Calculate the age of the wooden article. [$t_{1/2}$ of $C^{14} = 5460$ yr.]

4. Attempt the following : 15

(a) Explain energetics of β -decay.

(b) How is conservation observed in Nuclear reaction ?

Or

(a) What is nuclear reactor ? Describe its main parts briefly.

(b) What do you mean by Tracer technique ? Write different application of Tracer technique.

5. (A) Select the *correct* alternatives from the following : 5

(i) Radioactive isotopes that have an excessive neutron-proton ratio generally exhibit which one of the following :

(a) Alpha emission

(b) Beta emission

(c) Positron capture

(d) K-capture

- (ii) When hydrogen nuclei trap neutrons they form :
- Alpha particle
 - Deuterium
 - Beta rays
 - Positron
- (iii) Packing fraction is most closely related with :
- Dipole moment
 - Electron spin
 - Increase in relativistic mass
 - Mass defect
- (iv) The unit of effective cross-section in nuclear reaction is :
- 10^{-16} cm^2
 - 10^{-18} cm^2
 - 10^{-20} cm^2
 - 10^{-24} cm^2
- (v) The following isotope, the one that is not radioactive is :
- ${}^{60}_{90}\text{Co}$
 - ${}^{131}\text{I}$
 - ${}^{40}\text{Ca}$
 - ${}^{90}\text{Sr}$

(B) Write short notes on (any *two*) :

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- Buckshot Hypothesis
- Chemical Dosimeter
- Radioisotope as a source of electricity.