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AI—115—2017

FACULTY OF SCIENCE

M.Sc. (Fourth Semester) EXAMINATION OCTOBER/NOVEMBER, 2017

(CBCS Pattern)

ORGANIC CHEMISTRY

(CH-542/2)

(Bio-organic and Green Chemistry)

(Tuesday, 14-11-2017)

Time: 2.00 p.m. to 5.00 p.m.

Time— Three Hours

Maximum Marks—75

- N.B. := (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
 - (iii) Multiple choice questions (MCQ) should be attempted only once on page no. 3 of answer-book with complete answers.
- 1. Solve any three:

15

- (a) Role of PTC in Wittig's reaction.
- (b) Explain the structure of RNA.
- (c) Enzymes possess reaction specificity.
- (d) Atom economy of rearrangement reaction.
- (e) Use of DMC as a green reagent.
- 2. Answer the following (any three):

15

- (i) Explain nitrogen bases of Nucleic acid.
- (ii) Explain the following microwave solvent free reaction.
 - (a) Saponification of Ester;
 - (b) Alkylation of reactive methylene compounds.
- (iii) Microwave assisted Diel's-Alder reaction in organic solvent.
- (iv) Benefits of microwave in organic synthesis.
- (v) Explain induced fit mechanism for enzyme action.

P.T.O.

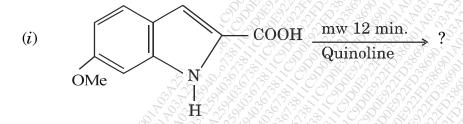
3. (A) What is green chemistry? Why is the need of green chemistry? Give the disadvantages of traditional method.

Or

Explain double helix structure of DNA.

(B) Predict the product (any four):

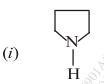
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$$(ii) \qquad \begin{array}{c} \text{OCOCH}_3 \\ \\ + \text{AlCl}_3 \end{array} \xrightarrow{?} \begin{array}{c} \text{OH} \\ \\ \text{COCH}_3 \end{array}$$

(iii)
$$+ H_3C \longrightarrow C \longrightarrow CH_3 \xrightarrow{K_2CO_3} ?$$

(iv)
$$\frac{\text{Pesudomonas}}{\text{Putida}}?$$



(v) Cyclohexanone

$$(ii) \operatorname{Br-CH}_2 - C \equiv C - \operatorname{CH}_3$$

$$(iii)$$
 H_3O

$$(vi) \qquad \begin{array}{c} OH \\ + CH_3OH \xrightarrow{MgO} ? \end{array}$$

4. (A) Explain three-point attachment theory for mechanism of enzyme action.

Or

What is atom economy? Explain atom economy in elimination and substitution reaction.

(B) Explain the applications of Ultrasound in chemical synthesis. 8

Or

Explain the role of ionic liquid in:

- (a) Reformatsky reaction;
- (b) Heck reaction;
- (c) Cannizaro's reaction;
- (d) Friedel-Crafts reaction.
- 5 (A) Select the correct alternative for the following multiple choice questions:
 - (i)RNA carries genetic information of specific protein synthesis in the form of code :
 - (a) r-RNA

(b) m-RNA

(c) t-RNA

(d) None of these

P.T.O.

WT		(4		AI—115—2017	
	(ii)	The concept of Atom Economy was developed by:			
		(a) Bary Trosf	(b) Joseph D.	Simon	
		(c) T. Remark	(d) None of the	nese	
	(iii)	used as PTC in Williamson's synthesis.			
		(a) K.tert. butoxide	K.tert. butoxide		
		(b) tetrabutyl ammonium salt			
		(c) tributyl-methyl ammonium salt			
		(d) Both (a) and (c)			
	(iv)	Alcohol dehydrogenase belongs toclass of			
		enzymes. (a) Hydrolase	(b) Transferas		
		(c) Oxireductase	(d) Lyase		
	(v)	is a green protocl for organic synthesis.			
		(a) Ultrasound	(b) Microwave	induced reaction	
	591.03	(c) Ionic liquids	(d) All of thes	e	
(B)	Write	short notes on (any two):			
	(i)	Classification and nome	enclature of enzymes;		

(ii)

(iii)

Factors affecting enzyme catalysed reactions;

Principles of green chemistry.