This question paper contains 3 printed pages]

## ST-454-2022

## FACULTY OF SCIENCE

## M.Sc. (First Year) (Second Semester) EXAMINATION MAY/JUNE, 2022

(New/CBCS Pattern)

**CHEMISTRY** 

Paper-CH-424

(Analytical Chemistry/Principles of Spectroscopy)

(Wednesday, 6-7-2022)

Time: 9.30 a.m. to 1.15 p.m.

Time— 3.45 Hours

Maximum Marks—75

N.B. := (i) Attempt all questions.

- (ii) Use of calculator or logarithmic table is allowed.
- (iii) Constants:

 $c = 3 \times 10^8 \text{ m/s}$ 

 $h = 6.626 \times 10^{-34} \text{ Js.}$ 

1. Attempt any *three* of the following:

15

- (a) Give an account of intensity of spectral lines.
- (b) The molecule  $H_2$  is microwave inactive while HCl is microwave active. Explain. Write a note on nuclear and electron spin interaction.
- (c) The pure rotational spectrum of the gaseous molecule CN consist of a series of equally spaced lines separated by 3.7978 cm<sup>-1</sup>. Calculate the internuclear distance of the molecule.

 $(^{12}C = 12.011 \text{ and } ^{14}N = 14.007 \text{ g.mol}^{-1}]$ 

- (d) How will you obtain photoelectron spectrum.
- (e) Give the use of NMR in medical diagnosis.
- 2. Answer any *three* of the following :

15

(a) Give an account of polarization and scattering of light radiation.

P.T.O.

Give the factors affecting the band position and intensities.

Explain Resonance Raman Sepctrum.

(*i*)

(ii)

WT (3) ST-454-2022

15

5. Write short notes on any *three* of the following:

- (a) Spin-spin interaction/coupling
- (b) Charge Transfer Spectra
- (c) Application of ESR Spectroscopy
- (d) Koopman's Theorem.

ST-454-2022

3