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**SB—03—2022**

**FACULTY OF SCIENCE**

**B.Sc. (Third Year) (Sixth Semester) EXAMINATION**

**JUNE/JULY, 2022**

**(CBCS/Old Pattern)**

**CHEMISTRY**

**Paper XIV**

**(Organic and Inorganic Chemistry)**

**(Thursday, 2-6-2022)**

**Time : 10.00 a.m. to 12.30 p.m.**

*Time—2½ Hours*

*Maximum Marks—40*

*N.B. :— (i) Attempt All questions.*

*(ii) Figures to the right indicate full marks.*

1. Answer any *three* of the following : 3×5=15

- (a) What are inner orbital complexes ? Explain with suitable example.
- (b) Define crystal field splitting ? Calculate CFSE for  $d^6$ ,  $d^7$ ,  $d^8$  and  $d^{10}$  configurations of tetrahedral complex.
- (c) Give the postulates of crystal field theory.
- (d) Write the different types of electronic transitions.
- (e) Discuss electronic spectrum of  $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$  complex ion.

2. Answer any *three* of the following : 3×5=15

- (a) How will you interpret IR spectra of the following compounds :
  - (i) Ethyne
  - (ii) 1-propanol
  - (iii) phenol.

P.T.O.

- (b) Define spin-spin splitting ? Predict the number of NMR signal of :
- Methanol
  - Diethyl ether
  - Acetone.
- (c) How will you synthesise  $\alpha$ -amino acid from strecker synthesis ? What is the action of  $\text{LiAlH}_4$  on glycine ?
- (d) Explain Favroskii rearrangement with mechanism.
- (e) The organic compound having molecular formula  $\text{C}_3\text{H}_5\text{N}$  shows the following spectral data :

UV—Transparent above 210 nm

IR—  $2975 \text{ cm}^{-1}$

$2210 \text{ cm}^{-1}$

PMR ( $\delta$  ppm) :  $\delta$  1.1, *t*, 3H

$\delta$  3.5, *q*, 2H

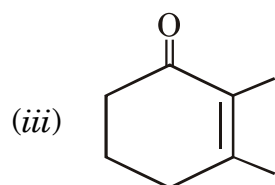
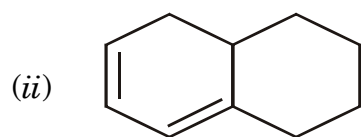
Deduce the structure and name of organic compound.

3. Answer any *two* of the following :

2×5=10

- Define chemical shift and give the advantages of tms ?
- Explain bathochromic and hypsochromic shift. Calculate  $\lambda_{\text{max}}$  of :





- (c) How will you synthesise peptide using DCC as reagent ?
- (d) Deduce the structure of compound based on the following PMR spectral data :

Molecular formula :  $C_2H_5Br$

PMR ( $\delta$  ppm) :  $\delta$  1.5, *t*, 3H

$\delta$  3.4, *q*, 2H