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**NA—01—2023**

**FACULTY OF SCIENCE**

**B.Sc. (Third Year) (Sixth Semester) EXAMINATION**

**NOVEMBER/DECEMBER, 2023**

**(New/CBCS Pattern)**

**CHEMISTRY**

**Paper-XIV**

**(Organic and Inorganic Chemistry)**

**(Wedday, 29-11-2023)**

**Time : 10.00 a.m. to 12.00 noon**

**Time—2 Hours**

**Maximum Marks—40**

**N.B. :—** (1) *All questions are compulsory.*

(2) *Figures to the right indicate full marks.*

1. Answer any *three* of the following : 3×5=15

- (a) Write the postulates of crystal field theory.
- (b) Define crystal field splitting. Calculate CFSE of  $d^3$  and  $d^6$  configuration in strong ligand field octahedral complex.
- (c) Explain John-Teller distortion effect in octahedral complex of  $\text{Cu}^{2+}$  ion.
- (d) Write a note on spectrochemical series.
- (e) Calculate ground state term symbol of  $d^3$  configuration.

2. Answer any *three* of the following : 3×5=15

- (a) Interpret IR spectrum of the following compounds :
  - (i) Ethane
  - (ii) 1-propanol
  - (iii) Phenol.

P.T.O.

- (b) Explain equivalent proton with example and predict number of NMR signal of :
- (i) Acetone
  - (ii) Ethylamine
  - (iii) Diethylether.
- (c) Define polymer. Explain free radical polymerization with mechanism.
- (d) Explain pinacol-pinacolone rearrangement reaction with mechanism.
- (e) The organic compound having molecular formula  $C_3H_8O$  shows the following spectral data :

UV : Transparent  $\lambda_{max} = 215 \text{ nm}$

IR :  $3600 - 3200 \text{ cm}^{-1}$

$2950 \text{ cm}^{-1}$

$1050 \text{ cm}^{-1}$

$^1\text{H-NMR}$  : ( $\delta$  ppm)

$\delta$  1.3, t, 3H

$\delta$  2.5, sextet, 2H

$\delta$  3.5, t, 2H

$\delta$  4.3, s, 1H

Deduce the structure of compound :

3. Answer any *two* of the following :

2×5=10

- (a) Explain different types of electronic transition.

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- (b) Explain shielding effect with a suitable example.
- (c) Give the synthesis and uses of Nylon-6, 10.
- (d) Deduce the structure of compound based on the following PMR spectral data molecular formula :  $C_7H_8$

PMR ( $\delta$  ppm) :  $\delta$  2.7, S, 3H

$\delta$  7.2–7.8, S, 5H

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